

August, 2015

STATE MEDICAL FACULTY OF WEST BENGAL

**Preliminary Examination For Diploma in Medical Laboratory Technology :
DMLT (Tech) Course**

PAPER – I : PATHOLOGY [Clinical Pathology & Haematology]

Time :2 Hours

Full Marks - 35

:GROUP A : ANSWER ANY TWO OUT OF THE FOLLOWING 3 QUESTIONS:- {2X10=20}

Q1. Mention different methods of E.S.R. estimation and describe the method most commonly used in a laboratory. What are the causes of increased E.S.R.?

3+4+3 = 10

Q2. Describe the process of semen analysis. State morphological abnormalities of sperm with labeled diagram. What is the range of normal sperm count?

5+4+1 = 10

Q3. How cell count and cell type are done in a sample of C.S.F.? How will you differentiate Tubercular and pyogenic meningitis by laboratory tests?

5+5 = 10

:GROUP B : WRITE SHORT NOTES ON ANY TWO OF THE FOLLOWING:-

{2X5=10}

- A. Reticulocyte count
- B. Bone marrow aspiration needle
- C. Occult blood test in urine.

:GROUP C : ANSWER ALL 5 QUESTIONS:-

{5X1=5}

1. Hypersegmented Neutrophil in blood smear is suggestive of:

- a) Iron deficiency
- b) Vitamin C deficiency
- c) Vitamin B12/Folic Acid deficiency
- d) Protein deficiency

2. Oxalate crystals are found in:

- a) Acidic Ph
- b) Alkaline Ph
- c) Neutral Ph
- d) None

3. Albumin in urine is detected by:

- a) Heat and Acetic acid test
- b) Heller's Nitric acid test
- c) Sulphosalicylic acid test
- d) All of them

4. Bilirubin in urine is increased in:

- a) Haemolytic jaundice
- b) Obstructive jaundice
- c) Both
- d) None

5. Basophil count is increased in:

- a) CML
- b) AML
- c) CLL
- d) ALL.

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**PAPER – II : MICROBIOLOGY
[General Bacteriology, Systemic Bacteriology, Immunology & Serology,
Clinical Bacteriology]**

Time : 2 Hours

Full Marks - 35

:GROUP A :

ANSWER ANY TWO OUT OF THE FOLLOWING 3 QUESTIONS:- {2X10=20}

- Q1. a) How will you prepare a smear from CSF for gram staining?
b) Write down the morphology of streptococcus pneumoniae and Neisseria meningitidis after gram staining with diagram.
4+6 = 10
- Q2. How will you –
a) perform the slide coagulase test?
b) Collect throat swab from a child?
c) Perform hanging drop preparation in the laboratory?
3+3+4 = 10
- Q3. How will you –
a) prepare oxidase reagent?
b) Perform oxidase test?
c) Do interpretation of oxidase test with example.
3+4+3 = 10

:GROUP B :

- Q4. Write Short Notes on: 2x5 = 10
A. RPR Test
B. Mantoux Test

:GROUP C :

- Q5. Answer the following:- 5x1 = 5
1. Clostridium tetani is sporing/non-sporing/capsulated/non-motile gram positive bacilli.
 2. Pressure of Autoclave is 15 pound per square inch/15 lb per square cm/15 kg per square inch/15 lb per square foot.
 3. Blood agar is enrichment/enriched/selective/basal media.
 4. Antibody which cross the placental barrier is IgG/IgM/IgA/IgD.
 5. BCG Vaccination is against Tuberculosis/Pertussis/Cholera/Diphtheria.

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**PAPER – III
BIOCHEMISTRY**

Time :3 Hours

Full Marks (Part-IIIA : 20 + Part-IIIB : 50)=70

Part IIIA and Part-IIIB are to be answered in separate Booklets

PART – III A

**[ELEMENTARY PRINCIPLES OF CHEMISTRY, PHYSICAL CHEMISTRY,
ORGANIC CHEMISTRY AND ELEMENTARY BIOCHEMISTRY]**

:-Attempt All Questions:-

- Q1. a) Write a note on Electrical and Chemical hazards in a laboratory. What measures (atleast two) must you take to avoid these two hazards? 2+2 = 4
- b) What do you mean by "10% Solution of Sucrose"? 2
- Q2. Define Buffer. State two suitable examples of Buffer. 2+2 = 4
- Q3. Write Short Notes on (**Any Three**):- 3x2 = 6
- a) Colloid and crystalloid
 - b) Reducing Sugar
 - c) Amino acid
 - d) DNA
- Q4. Write down the procedure of venous blood collection. 4

See overleaf

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**PAPER – III
BIOCHEMISTRY**

PART III B

[PRINCIPLES OF COMMON CLINICO-BIOCHEMICAL METHODS]

(Q1, Q2 & Q3 are Compulsory)

- Q1. Comment on the following statements (**True/False**): 5x1=5
- a) White Plastic bags are used in laboratory to dispose blood.
 - b) Rothera's test gives positive result if urine contains acetic acid.
 - c) Glycine is an amino acid.
 - d) Glucose is a normal constituent of urine.
 - e) Radio immuno assay uses radioactive isotopes.
- Q2. Write down the S.O.P. of a Centrifuge machine. What is hemolysis? How does it affect laboratory results? (atleast two examples) 6+2+2 = 10
- Q3. Write short notes on (**Any Three**):- 3 x5 = 15
- a. Standard Curve
 - b. Chromatography
 - c. pH Meter
 - d. Incubator
- (Answer any Two from Q4, Q5 & Q6)**
- Q4. Write in brief the procedure of Serum Electrophoresis. Draw a diagram showing electrophoretic separation of proteins. What are the functions of Serum Albumin (atleast two). 6+2+2=10
- Q5. Describe different parts of a Colorimeter. Name two instruments based on Colorimetric principle. Define Primary Standard Solution. 6+2+2= 10
- Q6. Describe the procedure of Serum Sodium and Potassium estimation. Mention normal range for both. What is hypokalemia? 6+2+2 = 10

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**PAPER – III
BIOCHEMISTRY**

PART III B

:PRINCIPLES OF COMMON CLINICO-BIOCHEMICAL METHODS.

(Q1, Q2 & Q3 are Compulsory and Answer any Two from Q4, Q5 & Q6)

Q1. Answer True or False (Overwriting will be taken as wrong answer): 5x2=10

- a) Sulphur is an important constituent of protein.
- b) Cellulose acetate paper is used for thin layer chromatography.
- c) Hyponatremia is a condition where plasma sodium is low.
- d) Apoenzyme + Holoenzyme = Coenzyme.
- e) Sucrose and lactose both are reducing sugar.

Q2. Write down the Standard Operating Procedure for Spectrophotometer and Electric Balance.

6+4 = 10

Q3. What are the preanalytic variations that can affect test result of blood glucose, blood urea and uric acid? How can a pipette of 1000ml be calibrated in the laboratory? What precaution should you take for drawing blood specimen for calcium?

6+2+2=10

Q4. What do you understand by the term "Arterial blood Gas"? How is an arterial sample sent? State with normal value the importance of pO₂, pCO₂ and pH.

1+3+6 = 10

Q5. Describe the tests to detect –

- a. Urinary glucose
- b. Ketone bodies in urine
- c. Blood in urine with their interpretation

10

Q6. How do you prepare standard curve of protein by Biuret method? Write with diagram the procedure of making a Standard Curve.

10
